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Throughout the text, spaces must be left between the last digits and storage units (10%, 20 wt.%, 30 ºС, 298 K). It is recommended to use a period as a decimal separator, and a short number "–" for the minus sign.

Mathematical formulas, chemical reactions are typed using built-in MS Word tool "Formula" either in Microsoft the equation is spanned by a narrow string (center alignment) and numbered (right in parentheses through taboo). The numbering is end-to-end and is given in the order of discovery, and only formulas (equations) that are referenced in the text are numbered. Structural formulas and complex chemical reactions can be inserted from ChemSketch or ChemDraw programs.

$С\_{p}\left(T\right)=3R\sum\_{i}^{}α\_{i}\frac{\left(\frac{Θ\_{i}}{T}\right)^{2}e^{\frac{Θ\_{i}}{T}}}{\left(e^{\frac{Θ\_{i}}{T}}-1\right)^{2}}$ (1)

SiF4 gas + 2 H2 gas + O2 gas → SiO2 TV + 4HF gas – 384 kJ (2)

Figures and plots are placed in separate paragraphs (text wrapping “on top and visible”, appearance 0 cm) and are accompanied by the appearance of a caption like “Fig. N . Title” (center alignment, no dot at the end). Add one empty line before and after the figure. Figures must be clear and have a resolution of 300-600 ppi (b/w or color). The file formats are tiff , jpeg or png .



Fig. 1. **A** DSC polymerization curves of American aryl cyanate in melt and solution at 4 K/min; **B** Dependence of activation energy on conversion for melt and solution polymerization



Scheme 1. Synthesis of galactofuranose glycosyl acceptors

Tables are supplied in as separate paragraphs, they are necessarily preceded by the name of the form “Table N. Title” (left-aligned, no dot at the end). Do not tint table headers, do not use bold font, use highlights only if it has a significant semantic load. Font size can be reduced to 11 pt. Add one empty line before and after the table.

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Table 1. Standard thermodynamic functions of silicon at 298.15 K

|  |  |  |  |  |
| --- | --- | --- | --- | --- |
|  | $$c\_{p}$$ | $$S°$$ | $$H°\left(298.15\right)-H°(0)$$ | Standard deviations (abs.) |
| $$c\_{p}$$ | $$S°$$ | $$H°\left(298.15\right)-H°(0)$$ |
| J mol –1 K –1 | J mol –1 | J mol –1 K –1 | J mol –1 |
| Nast. Job | 20.034 | 18.794 | 3214.6 | 0.0 12 | 0.0 16 | 1.4 |
| NIST-JANAF [3] | 20.00 0 | 18.82 0 | 3217.9 | — | — | — |

*Acknowledgments, grant support and sources of funding are gradually accumulated at the end of the abstract text in a separate paragraph before the references.*

**References**

The list of references is given numbered in the order in which the work is mentioned in the text of the abstract and is drawn up in accordance with GOST R 7.0.5-2008. For all sources in the text of the abstract, a reference to the type of number in square brackets [1,2] is required.

1. Cai J.-K., Chen Y.-P. Application of the Peng-Robinson Equation of State with Volume Transformation to Vapor-Liquid Equilibrium Calculations // Fluid Phase Equilib. 1998. Vol. 145. S. 193-215.

2. CRC Handbook of Chemistry and Physics. 102nd ed. / ed. Rumble J.R. Boca Raton, FL: CRC Press, 2021.